ANURAG Engineering College

(An Autonomous Institution)

I B.Tech I Semester Supplementary Examinations, June/July-2024

ENGINEERING CHEMISTRY (COMMON TO CIVIL, EEE & MECH)

Time: 3 Hours		Max. Marks: 75		
Section – A (Short Answer type questions) Answer All Questions		Course	(25 B.T	Marks) Marks
Allswe	All Questions	Outcome	Level	
1.	Explain atomic orbital and molecular orbital?	CO1	L1	2M
2.	Draw MOED diagram of O ₂	CO1	L2	3M
3.	What is scale and sludge?	CO2	L1	2M
4.	Write about caustic embrittlement?	CO2	L2	3M
5.	What is electrode potential?	CO3	L1	2M
6.	Summarize normal hydrogen electrode.	CO3	L2	3M
7.	What is anti-Markonikoff's rule?	CO4	L1	2M
8.	Define terms, enatiomer and diastereomer.	CO4	L1	3M
9.	How to prepare Nylon-6,6?	CO5	L1	2M
10.	What are the applications of biodegradable polymers?	CO5	L2	3M
	Section B (Essay Questions)			
Answer all questions, each question carries equal marks.		(5	X 10M	=50M)
11. A)	Write about LCAO method.	CO1	L3	10M
11.11)	OR			
B)	Write crystal field splitting patterns of metal ion d-orbital of i) Tetrahedral geometries ii) Octahedron geometries	CO1	L3	10M
12. A)	i) What is the theme of desalination of brackish water by reverse osmosis method.	CO2	L3	5M
	ii) Distinguish between temporary hardness and permanent hardness.			5M
	OR			
B)	Construct the ion exchange method for softening of water with neat diagram.	CO2	L3	10M
13. A)	i) How to determine the P ^H of the solution by calomel electrode. ii) Explain metallic coating by electroplating of copper.	CO3	L3	5M 5M
B)	What is corrosion. Classify the chemical and electro-chemical corrosion.	CO3	L3	10M
14. A)	Write preparation method, uses and side effects of Asprin. OR	CO4	L3	10M
B)	Explain i) Oxidation of alcohols by KMnO ₄ and CrO ₃ ii) Reduction of carbonyl compounds by LiAlH ₄ and NaBH ₄	CO4	L3	10M

R18

Question Paper Code: R18A11HS02

15. A)	Distinguish between thermoplastics and thermosets.	CO5	L3	10M
B)	OR			10111
	Write structure, properties and applications of poly glycolic and poly lactic acid.	CO5	L3	10M